

CHAPTER 1

INTRODUCTION: *A New Continental Perspective on an Old and Vexing Problem*

The North American continent is a vast industrialized landmass governed by three sovereign nations with distinct cultures. The atmosphere that blankets Canada, the United States, and Mexico, however, is a common resource and its protection is a shared responsibility. Ground-level ozone (O_3) is an atmospheric pollutant (see Textbox 1.1 and Figure 1.1) that poses a pervasive air-quality problem throughout the continent. Its detrimental effects on human health and ecosystems (cf. Federal Working Group, 1999; U.S. EPA, 1996; DDF, 1997) disregard national borders and its management has proven to be complex. What are the processes that cause elevated concentrations of ground-level O_3 in North America? How can our understanding of these processes be used most effectively to manage and

reduce the adverse effects of tropospheric O_3 ? These are the questions that form the focus of this report.

Commissioned by NARSTO,^a this state-of-science assessment provides a unique tri-national perspective on the ground-level O_3 issue. Its goals are to document scientific efforts to explain how O_3 is formed and transported, as well as to describe policies intended to control and mitigate damage. Its intent is to build upon knowledge and past success to improve management of O_3 pollution across the continent. Because scientists and policy makers are partners in the effort to successfully balance industrial progress and environmental protection, this document is intended to serve both groups as an audience.



1.1 The Tropospheric Ozone Pollution Problem

Despite its seemingly minute concentration, atmospheric O_3 has an enormous environmental impact. In the stratosphere, where about 90% of the atmosphere's O_3 resides, it protects life from harmful ultraviolet radiation. In the troposphere, O_3 is a key part of an oxidizing system that cleanses the atmosphere of a wide range of pollutants. At ground level, however, O_3 can damage human health, agricultural yields, and other ecosystems.

Ground-level O_3 abundances in non-urban regions of North America typically range from 20 to 70 ppb^a (LeFohn et al., 1998). Hourly averaged values in excess of 100 ppb can occur, however, during regional or non-urban air pollution episodes. Peak O_3 levels during urban air-pollution episodes tend to be substantially higher, with peak hourly-averaged values often exceeding 100 ppb and sometimes exceeding 200 ppb. The highest hourly-averaged O_3 mixing ratio on record in North America is 680 ppb, measured in downtown Los Angeles in 1955. When O_3 concentrations reach these levels, they impair lung function in humans and other animals and inflict foliar damage to agricultural crops, forests, and other vegetation (e.g., reducing yields and net primary production).

^a Throughout this report we refer to O_3 abundances in terms of "mixing ratios," where 1 ppb O_3 signifies a ratio of 1 molecule of O_3 to every billion molecules of air and 1 ppm a ratio of 1 molecule of O_3 to every million molecules of air.

^aFormerly an acronym for "North American Research Strategy for Tropospheric Ozone," the term NARSTO has become simply a wordmark signifying a tri-national, public/private partnership for dealing with multiple features of tropospheric pollution, including O_3 and suspended particulate matter. Additional information on NARSTO is available on <http://www.cgenv.com/Narsto>.

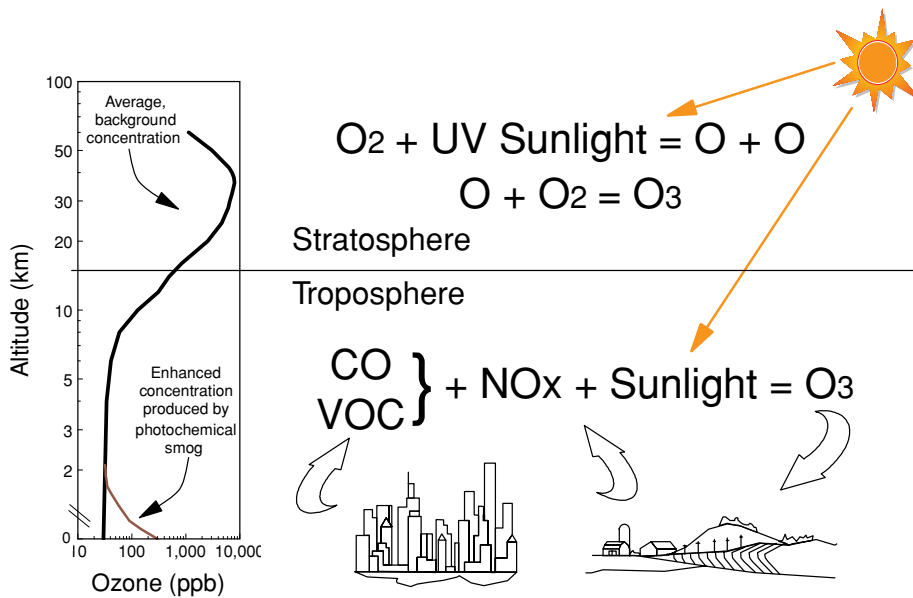


Figure 1.1 Atmospheric ozone (O₃). At left: The mixing ratio of atmospheric O₃ as a function of height. At right: Photochemical processes responsible for producing O₃ in the stratosphere and troposphere (including at ground level). VOC = volatile organic compound; NO_x = NO + NO₂ = nitrogen oxide; UV = ultraviolet.

1.1 THE MOTIVATION - Why a Scientific Assessment Now?

This NARSTO assessment represents one in a sequence of assessments of ground-level O₃ pollution in North America spanning more than two decades. The current assessment has been undertaken for important reasons:

- The problem persists.
- New and emerging air-quality policies create a need for scientific input.
- New scientific findings have implications for policy.
- There is a need for a continental perspective on ozone management.

1.1.1 The Problem Persists

Emission controls to mitigate O₃ pollution began to be implemented in the United States and Canada in the 1970s and in Mexico in the 1980s (see Textbox 1.2 and Chapter 2). Trends in O₃ concentrations at various urban locales in North America over the past decade suggest that these controls have produced beneficial results. For example, in the United States, O₃ concentrations have generally been headed downward (Figure 1.3). Of

special note are the trends illustrated in Figure 1.4. These indicate that, *despite major growth in population and economic activity*, peak O₃ concentrations declined significantly in Los Angeles (where O₃ pollution has historically been the most severe in North America) and have been relatively constant in other particularly high-impact areas such as Houston and Mexico City.

However, the problem of ground-level O₃ pollution in North America is far from solved. In 1995, it is estimated that some 70 million U.S. citizens, over 13 million Canadians, and 20 million Mexicans lived in or near areas where O₃ concentrations violated O₃ exceedance levels (U.S. EPA, 1996b; Canadian-Summary, 1997; SEMARNAP 1997). Moreover, 27 of 29 urban areas required by the U.S. Clean Air Act Amendments of 1990 to submit State Implementation Plans (SIPs) were unable to meet the 1994 deadline (see Section 2.2.4.1) because of the difficulties associated with developing a workable plan to reach attainment. The persistence of the O₃ pollution problem in the face of concerted, sustained, and often costly efforts to address it has raised a perplexing question for both the scientific and policy-making communities: *To what extent does our inability to mitigate O₃ pollution in North America reflect an inadequate scientific understanding of the underlying*

causes of the phenomenon, a faulty application of this understanding, and/or simply the inability of society to control the socioeconomic drivers of O₃ pollution? (see Textbox 1.3). Addressing this question has proven to be difficult, in large part because of a lack of sufficient monitoring over the past two decades. One way of approaching this issue would be to integrate an iterative tracking system into O₃ management programs (see Chapter 5).

1.1.2 New and Emerging O₃ and Particulate Policies Create Need for Scientific Input

1.1.2.1 Revised Ambient O₃ Standards and Goals

Recent medical studies have uncovered statistically significant relationships between adverse health effects in humans (e.g., as measured by hospital admissions) and ambient O₃ mixing ratios well below 100 ppb (see, for example, Canadian-Summary, 1997 and the U.S. EPA, 1996a, and the references cited therein). These studies have precipitated a major reappraisal of the air-quality goals, standards, and policies of the North American nations related to O₃. In the United States, the NAAQS has been lowered from a 1-hr, 0.12-ppm standard to an 8-hr, 0.08-ppm standard.^{b,c} In Canada, an assessment designed to develop a new O₃ objective is currently underway.

While a critical review of the underlying health-effects data driving these new policies is beyond the scope of this assessment, the implications of these policies to the atmosphere are highly relevant. Our understanding of tropospheric O₃ suggests that attainment of these new goals and standards will not be a trivial matter and will require significant changes in the methods and approaches used to meet them. For example, as O₃ goals and standards are shifted to lower concentrations, it is likely that control strategies

will have to further emphasize the already emerging focus on regional, or even continental emissions (see Section 3.8).

1.1.2.2 New Ambient Standards and Goals for Fine Particles

The fine-particle category commonly referred to as PM_{2.5} includes atmospheric particulate matter with aerodynamic diameters less than 2.5 μm.^d Because of their small size, such particles can penetrate into the deep lungs of a respiring individual and cause respiratory and other health effects. Recent epidemiological studies (NRC, 1998) have been interpreted to indicate a statistical association between adverse health effects and relatively low concentrations of PM_{2.5}. As a result, a new NAAQS for PM_{2.5} has been promulgated in the United States (Federal Register, 1997). The U.S. NAAQS for PM₁₀, defined as particles with aerodynamic diameters less than 10 μm, remains in effect. In Canada, a scientific review and evaluation of atmospheric particulate matter was undertaken in 1997. New health-based air-quality objectives have been recommended for PM_{2.5} as well as PM₁₀.

As is the case for O₃, a critical assessment of epidemiological data on the health effects of particulate matter is beyond the scope of this report. However, the technical issues surrounding the mitigation of O₃ pollution and PM_{2.5} are interdependent. O₃ and PM_{2.5} are coupled through common photochemical pathways and common precursors, as shown in Figure 1.5 and have similar dependencies upon meteorology during the warm months. Thus, control strategies designed to address one pollutant can affect the other. As discussed in Section 3.8, a major challenge for the coming decades will be to develop control strategies that take advantage of these interdependencies.

^b The NAAQS are stated in units of ppm and to two significant figures. Thus, an exceedance is technically defined as a concentration of at least 0.125 and 0.085 ppm, respectively; i.e., concentrations that when rounded to two significant figures yield values equal to or greater than 0.13 and 0.9 ppm.).

^c At the time this report was written, these revised standards were the subject of ongoing litigation. Regardless of the outcome of this litigation, the development and promulgation of these standards form part of the historical context in which this report was developed.

^d The regulatory definition of PM_{2.5} is “particles that behave aerodynamically as if they had diameters of 2.5 μm or less.”



1.2 Historical Background

By the early 1960s, the contributors to O_3 pollution appeared to have been identified (as shown in the timeline of Table 1.1). Ground-level O_3 pollution is produced by the oxidation of VOCs in the presence of sunlight and NO_x , as shown in Figure 1.1. VOC is derived from motor vehicle exhaust, evaporation of commercial and industrial solvents, and fugitive emissions from industry, while NO_x is emitted as a by-product of combustion in motor vehicles, power plants, and other stationary sources. In the presence of these sources and under the appropriate meteorological conditions (i.e., high temperatures, low wind speeds), photochemical reactions quickly generate the large O_3 concentrations typically observed during urban O_3 pollution episodes. Control of ground-level O_3 pollution in an urban area can thus be accomplished, in principle, by simply reducing the local emissions of VOC and/or NO_x . With this base of scientific understanding, and medical and epidemiological studies documenting the adverse health effects of O_3 , the stage was set for the promulgation of policies designed to address the problem. The United States, through the Clean Air Act of 1970, was the first of the three nations of North America to adopt a national air-quality goal for O_3 (referred to in the United States as a National Ambient Air Quality Standard or NAAQS) and a federally coordinated program to reach attainment of this goal within specific deadlines. Soon after, Canada and Mexico implemented their own control policies. On the basis of relatively primitive analyses of smog chamber and air-quality data, it was concluded that reductions in urban VOC emissions would provide the most effective strategy for reducing O_3 , and, as a result, initial O_3 abatement strategies tended to focus on the implementation of *local* VOC emission controls complemented by nationwide controls on automotive emissions of NO_x as well as VOC (see Figure 1.2).

Table 1.1 The Birth of Science of O_3 Pollution and Its Control

<u>Year</u>	<u>Milestone</u>	<u>Notes</u>
1840s - 50s	O_3 molecule discovered O_3 presence in atmosphere documented	Schoenbein (1840a&b, 1854)
1874	O_3 shown to be toxic to animals	Andrews (1874)
1940s	Photochemical smog found to be causing crop damage	Middleton (1950)
1950s	O_3 found to be major oxidant in photochemical smog VOC and NO_x shown to be O_3 photochemical precursors	Haagen-Smit (1952)
1961	Basic science of O_3 pollution documented in monograph	Leighton (1961)
1970	U.S. Clean Air Act of 1970 establishes first national program for the mitigation of O_3 pollution in North America	
1971	Canada establishes O_3 air-quality objective	
1970s -80s	Initial O_3 mitigation strategies tend to emphasize local VOC controls	see Figure 1.2

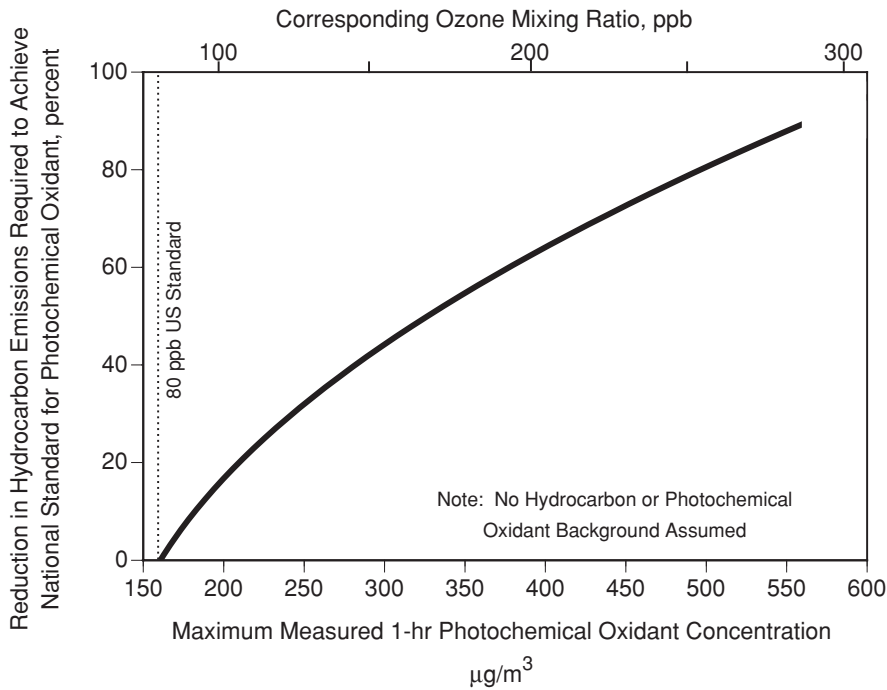
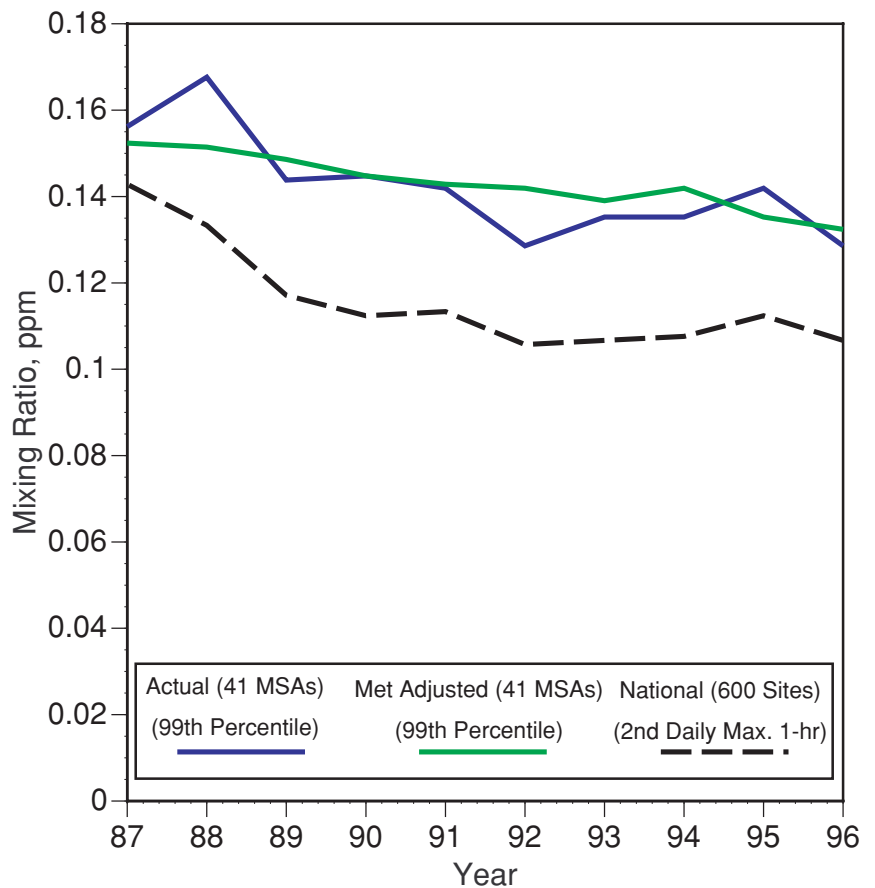


Figure 1.2 Initial guidance from the U.S. EPA on the required reduction in hydrocarbon emissions needed to achieve the NAAQS for photochemical oxidants (at that time of 80 ppb or $160 \mu\text{g m}^{-3}$) as a function of the measured peak photochemical oxidant concentration. Current scientific understanding suggests that a significantly more complex situation often applies that will likely require a combination of VOC and NO_x controls (see Section 3.2) Source: Federal Register, V. 36, No. 158, 1971.

Figure 1.3 Actual and meteorologically adjusted composite average O_3 trends at 41 U.S. Metropolitan Statistical Areas (MSAs) and 600 urban, suburban, and rural sites. After U.S. EPA, 1997.



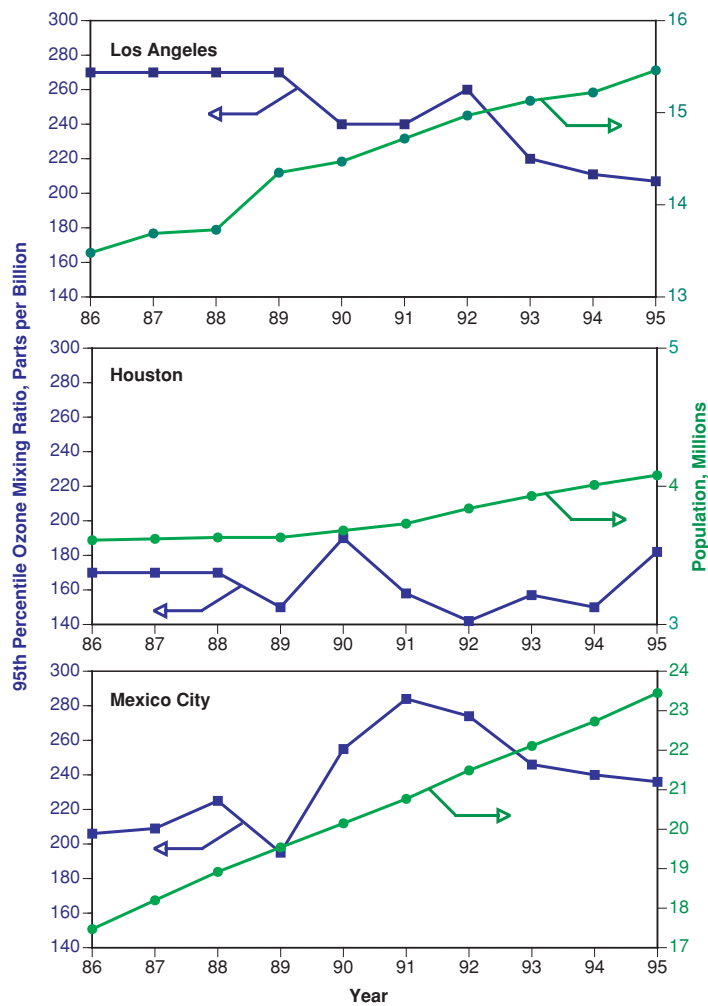


Figure 1.4 Trends in annual 95th percentile daily O₃ maxima (blue) and population (green) at Los Angeles (top), Houston (middle), and Mexico City (bottom). O₃ data obtained from the U.S. Aerometric Information Retrieval System (AIRS) (URL [http://capita.wustl.edu/otag/DATA/AIRS/Data.HTML](http://capita.wustl.edu/otag/DATA/AIRS>Data.HTML)) for Los Angeles and Houston and from SEMARNAP (1997) for Mexico City. Population data were obtained from standard census sources and reflect populations in the Metropolitan Statistical Area of each city.^c

1.1.3 New Scientific Findings Have Policy Implications

Scientists continue to significantly improve their understanding of the complex phenomena that cause O₃ pollution. For this improved understanding to be implemented, the conceptual framework and tools used by the policy maker in addressing O₃ abatement must also evolve. Improved communication and technology transfer at this science-policy interface are critically important to informed air-quality management decisions. New scientific insights into the complexity of the O₃

pollution phenomenon already have begun to spark significant changes in the policies adopted for managing O₃ pollution in North America; in turn, these have placed new burdens upon the scientific and technical communities. Key aspects of this changing science/policy landscape relate to O₃ trends, VOC and NO_x emission controls, emission inventories, biogenic precursors, meteorological processes, measurement technologies and networks, new observation-based methods, and the implications of new air-quality standards and objectives. These aspects are discussed in Chapter 3.

^cThese urban population increases may be compared with national population growth rates between 1900 and 2000 as follows: Canada -- 27.8M - 31.3M (12.6%); U.S. -- 250.0M - 275.6M (10.2%); Mexico -- 84.5M - 100.4M (18.8%).

i 1.3 The Role of Science in Addressing the Ozone Pollution Problem

As with most environmental problems, **scientific understanding is an essential but not the sole determinant** in the mitigation of ground-level O_3 pollution. Other factors related to technology, economics, and societal priorities can be equally if not more critical. Indeed, in the final analysis our scientific understanding of O_3 pollution is already adequate to alleviate the problem. It is virtually certain that ground-level O_3 pollution is caused by photochemical reactions involving the oxidation of VOC and CO in the presence of NO_x (see Textbox 1.1), and thus, in principle, society's contribution to ground-level O_3 could be eliminated by ceasing all anthropogenic emissions of VOC, CO, and NO_x . For obvious reasons, such a solution to the O_3 pollution problem is not currently a viable option. Society has instead opted to attempt to bring O_3 concentrations down to acceptable levels of risk (i.e., levels that minimize adverse effects on human health and welfare) without unduly disrupting the technological infrastructure that underpins our economy and unprecedented standard of living. As a result, the scientific community has been given a much more difficult task than simply elucidating the causes of the pollution. It must identify the mix of VOC, CO, and NO_x emission reductions that will be most effective in lowering O_3 concentrations and determine the minimum amount of emission reductions necessary to meet a given O_3 air quality standard or objective. Actual mitigation of O_3 pollution then requires the conversion of this scientific understanding into actual emission reductions through the promulgation of rules and regulations on specific types of emissions and the development and implementation of emission control technologies to comply with these rules and regulations. Economic and demographic projections also play a critical role, since economic and/or population growth can offset the emission reductions achieved through the application of an emission control technology. Efforts to mitigate O_3 pollution can be hampered by a breakdown in any component of this process.

Oxidant Production and Fine Particle Production

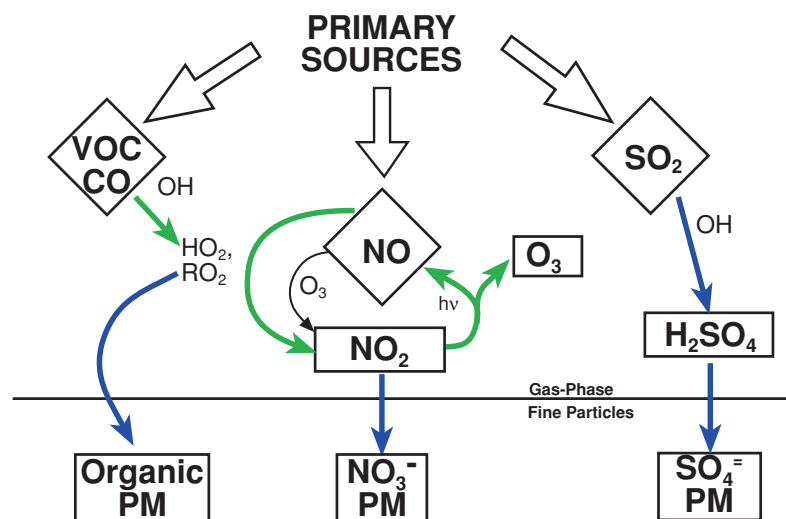


Figure 1.5 Schematic illustrating coupling between the sources of primary gaseous pollutants (diamonds), the photochemical processes leading to production of O_3 (green arrows) and those leading to the formation of fine particulate matter containing organics, nitrates, and sulfates (blue arrows).

CHAPTER 1

1.1.4 Need for Continental Perspective in Policy Making

1.1.4.1 The Regional Perspective

In the 1960s, photochemical smog was perceived as a basically urban problem. As a result of this perception, measures to address O₃ pollution focused mainly on local or urban-scale emissions. For example, in the United States, a two-level approach has been adopted: 1) nationwide controls on mobile and certain other sources; and 2) controls implemented on local, usually urban, scales. Determination of the final mix of controls implemented to reach the U.S. NAAQS has been in the past, and remains today, the responsibility of state and local authorities within a given non-attainment area. Similarly, design and implementation of measures to reach the ambient air quality objective in Canada and Mexico are a shared responsibility of the federal, provincial and local governments, comparable to the two-level system in the United States.

Despite the historical emphasis on local emission controls and the implementation of several federal control programs, observations in rural areas of North America show O₃ pollution to be ubiquitous to much of the North American continent. Observations in the 1970s indicated that photochemical smog can be regional in its spatial extent (i.e., 1000s of kilometers) and, given appropriate meteorological and topographic conditions, it can produce elevated concentrations of ground-level O₃ over rural as well as metropolitan areas (e.g., RTI, 1975; Decker, 1976; Wolff and Liroy, 1980). Data from recently established rural O₃ monitoring networks in the northeastern and southeastern United States, in California, and in portions of southern Canada have confirmed that rural O₃ pollution episodes do occur with 1-hr peak O₃ concentrations typically from 60 to 100 ppb (Saylor et al., 1998; Canadian - Vegetation, 1997). Figure 1.6 indicates the accumulated O₃ dosage over the eastern United States during the growing season of 1995. Analysis of these and similar rural O₃ observations in conjunction with empirically derived dose-response relationships between ambient O₃ and vegetation suggests that

these regional-scale episodes are severe enough, and recur often enough over a growing season, to have significant effects on the net productivity of agricultural crops, forests, and other ecosystems in portions of the North American continent (Heck and Cowling, 1997; Canadian - Vegetation, 1997).

The realization that O₃ pollution is regional in nature and can occur in rural as well as urban areas has sparked a reappraisal of policies and scientific understanding in North America. On the policy side this reappraisal has led to new initiatives to develop regional approaches to O₃ management (e.g., the Ozone Transport Assessment Group, discussed in Section 2.2.1). On the scientific and technical side, this reappraisal has led to a renewed effort to understand the mechanisms that cause long-range transport and urban-rural interchange. In turn, a reevaluation of the instrumentation, measurement strategies, and models used by the scientific community and their ability to characterize the processes that couple urban and rural pollution (see Section 3.6 and Chapter 4) is underway. This scientific reevaluation has provided a deeper understanding of the roles of VOC and NO_x in fostering O₃ pollution and how these roles tend to evolve in time and space (see Chapter 3).

1.1.4.2 A Potentially Rising “Continental Background” Raises Additional Concerns

There is now a growing body of evidence that suggests that the spread of O₃ pollution may extend beyond the regional scale to the continental, and even hemispheric scale. Detailed analyses of late 19th and early 20th century O₃ datasets indicate that free tropospheric O₃ in the northern hemisphere has undergone significant variations over the past century and continues to vary in some regions of the globe today. (e.g., Logan, 1988; Volz and Kley, 1988; Staehlein et al., 1994; Scheel et al. 1997). The evidence for long-term variations in free tropospheric O₃ is strongest in Europe where measurements of O₃ date back to the 1800s. For example, Volz and Kley (1988) used current techniques for measuring O₃ to recalibrate O₃ measurements made in Paris in the late 19th century. Their analysis indicates that surface O₃ concentrations at the end of the 1800s were about a

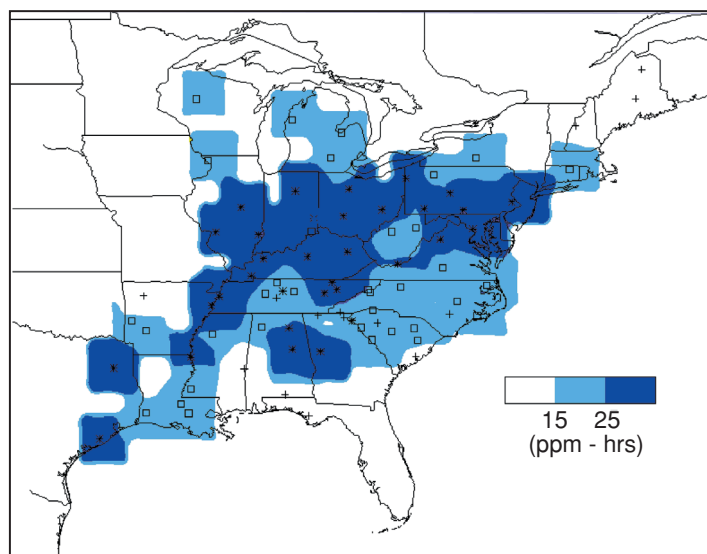


Figure 1.6 Growing-season exposure of crops and other vegetation in the eastern United States to O_3 in 1995 in terms of the crop-exposure indice “SUM06” (in ppm-hrs). The larger the SUM06 value the greater the potential damage to vegetation and the greater the probability crop yields will be depressed. Data from the U.S. National Crop Loss Assessment Network indicate that agricultural yields in North America are depressed, on average, by about 10% when SUM06-values exceed 15-25 ppm-hrs (Heck and Cowling, 1997, Canadian - Vegetation, 1997). SUM06 values shown here were calculated from an analysis of 85 rural O_3 monitoring sites that comprise the SOS, SON, and EPA CASTNet networks. Locations of sites are indicated by: “+” for sites with SUM06 < 15 ppm-hrs, “box” for sites with 15 < SUM06 < 25 ppm-hrs, and “*” for sites with SUM06 > 25 ppm-hrs. (After Saylor et al., 1998.)

factor of 2 less than current concentrations over Europe. More detailed analyses following Volz and Kley’s seminal paper have reached similar conclusions (see, for example, Figure 1.7).

Because there are no records of systematic O_3 concentration measurements over North America before the mid-20th century, it is not possible to determine if a similar increase in O_3 over North America has occurred. Nevertheless, the similarities between Europe and North America in terms of land use and land cover, pollutant emissions, and meteorology suggest that long-term trends in O_3 over North America may be like those inferred for Europe. The implications of such a long-term O_3 trend for understanding O_3 pollution effects on humans and ecosystems and for possible future approaches to mitigate these effects have yet to be fully explored but could be considerable (see for example, Sections 3.1 and 3.8.)

1.2 REPORT STRUCTURE AND PRESENTATION – *What’s Here*

The audience for this assessment reflects a broad spectrum of policy-relevant and technical expertise. In view of this spectrum, the presentation is structured in a manner that addresses the needs of both disciplines. Chapter 2 provides an overview of the O_3 management approaches that have been adopted by Canada, Mexico, and the United States. Chapter 3 summarizes key findings in the science of O_3 pollution with specific emphasis on their relevance to the policy-making community. Chapter 4 focuses on Air-Quality Modeling Systems: their characteristics, recent improvements, and inherent limitations. Chapter 5 describes how the tools of the scientific community can be integrated into an O_3 management scheme that is robust from a scientific point of view and accountable and responsive from a policy point of view. Chapter 6 concludes the assessment with a a brief summary of major conclusions and recommendations for future directions.

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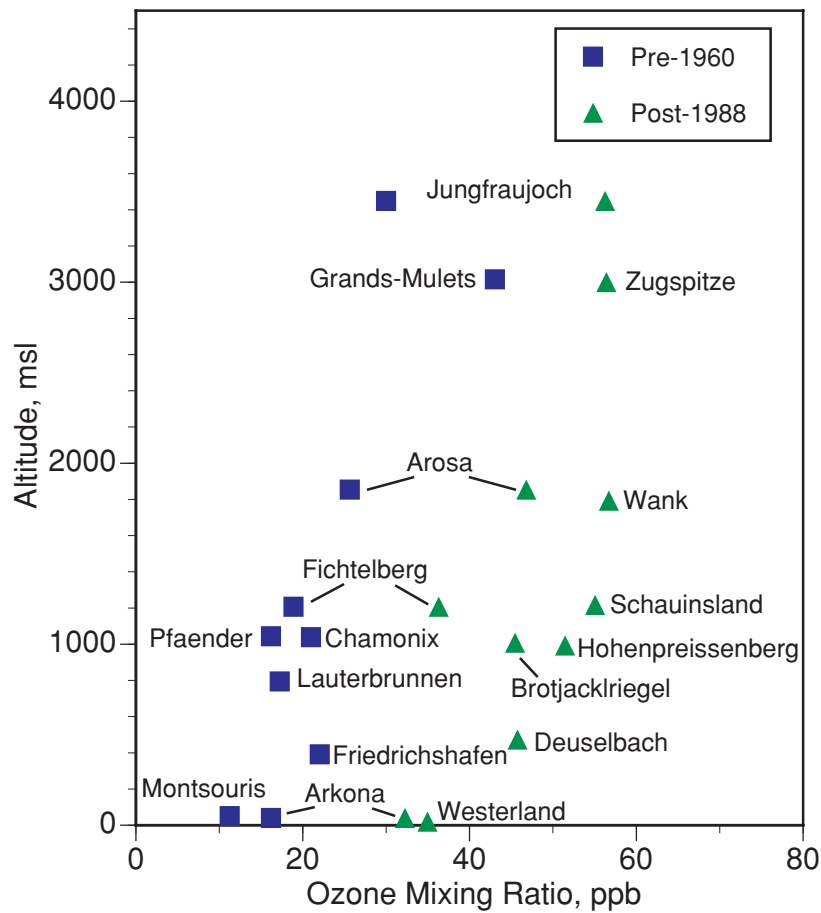




Figure 1.7 Surface O₃ concentrations of August/September from different locations in Europe as a function of altitude. Squares indicate averaged O₃ concentrations measured between 1896 (Grand Mulets) and 1940 (Pfaender) and the triangles indicate averaged O₃ concentrations measured between 1988 and 1991. (After Staehlein et al., 1994.)

Throughout the document, textboxes are used to highlight information. These boxes contain informational or technical material and are marked accordingly:

-  indicates information of general interest (blue background)
-  indicates technical details for the interested reader (green background).